

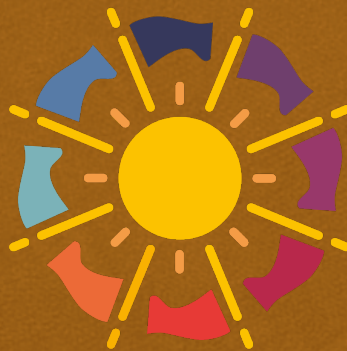
PHYS 3038 Optics

L4 EM Theory, Photons, & Light

Reading Material: Ch3.3-3.7



Shengwang Du



2015, the Year of Light

3.3 Energy & Momentum

Vacuum



$$\left. \begin{aligned} u_E &= \frac{\epsilon_0}{2} E^2 & u_B &= \frac{1}{2\mu_0} B^2 \end{aligned} \right\} \quad u = u_E + u_B = \epsilon_0 E^2 = B^2 / \mu_0 \quad \text{Energy density}$$

For plane wave: $E = CB$

Energy flow density $S = uc = c\epsilon_0 E^2 = c\epsilon_0 EcB = c^2\epsilon_0 EB = \frac{1}{\mu_0} EB$

Poynting vector $\vec{S} = c^2\epsilon_0 \vec{E} \times \vec{B} = \frac{1}{\mu_0} \vec{E} \times \vec{B}$

Medium

$$u = \frac{1}{2}(\vec{E} \cdot \vec{D} + \vec{H} \cdot \vec{B}) \quad \vec{S} = \vec{E} \times \vec{H}$$

Average Energy and Power of EM Wave

$$\vec{E} = \vec{E}_0 \cos(\vec{k} \cdot \vec{r} - \omega t) = \text{Re}\{e^{i(\vec{k} \cdot \vec{r} - \omega t)}\}$$

Optical frequency $\sim 10^{15}$ Hz



$$\begin{aligned} u &= \epsilon_0 E^2 = \epsilon_0 E_0^2 \cos^2(\vec{k} \cdot \vec{r} - \omega t) = \frac{1}{2} \epsilon_0 E_0^2 [1 + \cos 2(\vec{k} \cdot \vec{r} - \omega t)] \\ &= \frac{1}{2} \epsilon_0 E_0^2 + \frac{1}{2} \epsilon_0 E_0^2 \cos 2(\vec{k} \cdot \vec{r} - \omega t) \end{aligned}$$

Time-averaged $\langle u \rangle_T = \frac{1}{2} \epsilon_0 E_0^2 + \frac{1}{2} \epsilon_0 E_0^2 \langle \cos 2(\vec{k} \cdot \vec{r} - \omega t) \rangle_T = \frac{1}{2} \epsilon_0 E_0^2$

Intensity $I = \langle S \rangle_T = c \langle u \rangle_T = \frac{1}{2} c \epsilon_0 E_0^2$ average energy per unit area and per unit time (W/m²)

Power $P = IA = \frac{1}{2} c \epsilon_0 E_0^2 A$ average energy through the area A per unit time (W)

3.3.3 Photons

☞ Photon: particle (quantization) of EM field – chargeless and massless



Photon of a plane wave $(\vec{k}, \omega)$

Energy of a photon in the plane wave: $\hbar\omega = h\nu$

Momentum of a photon in the plane wave: $\hbar k = \frac{h}{\lambda}$

Mean photon flux $\Phi = \frac{P}{\hbar\omega}$

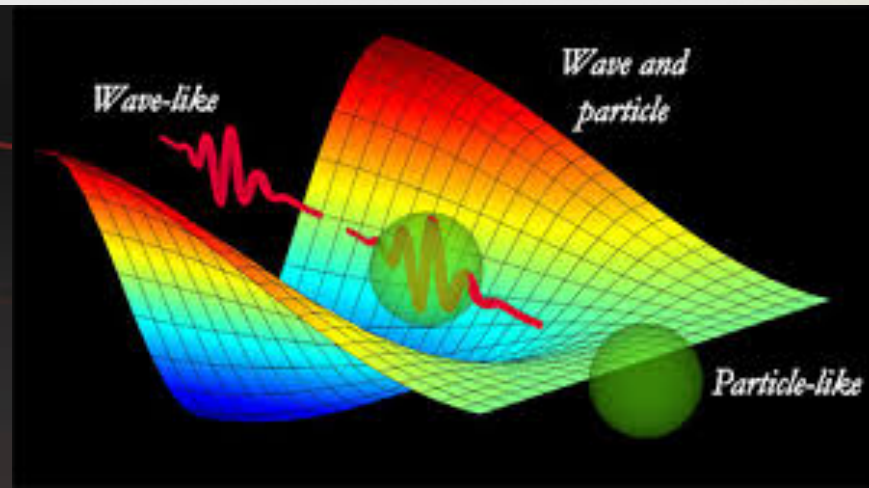
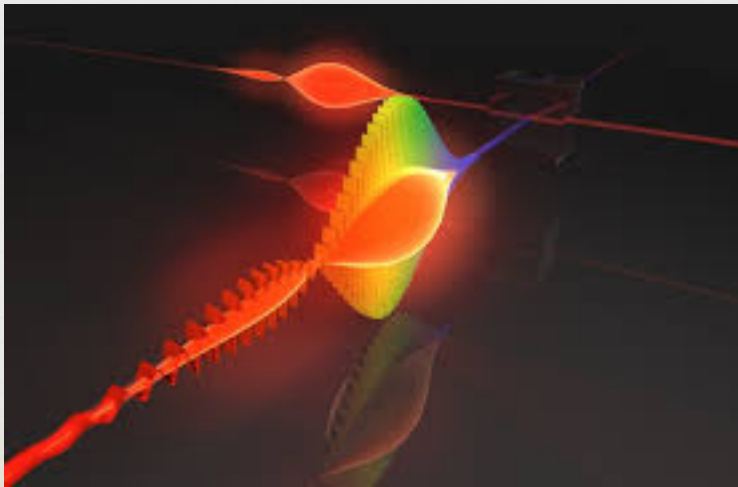
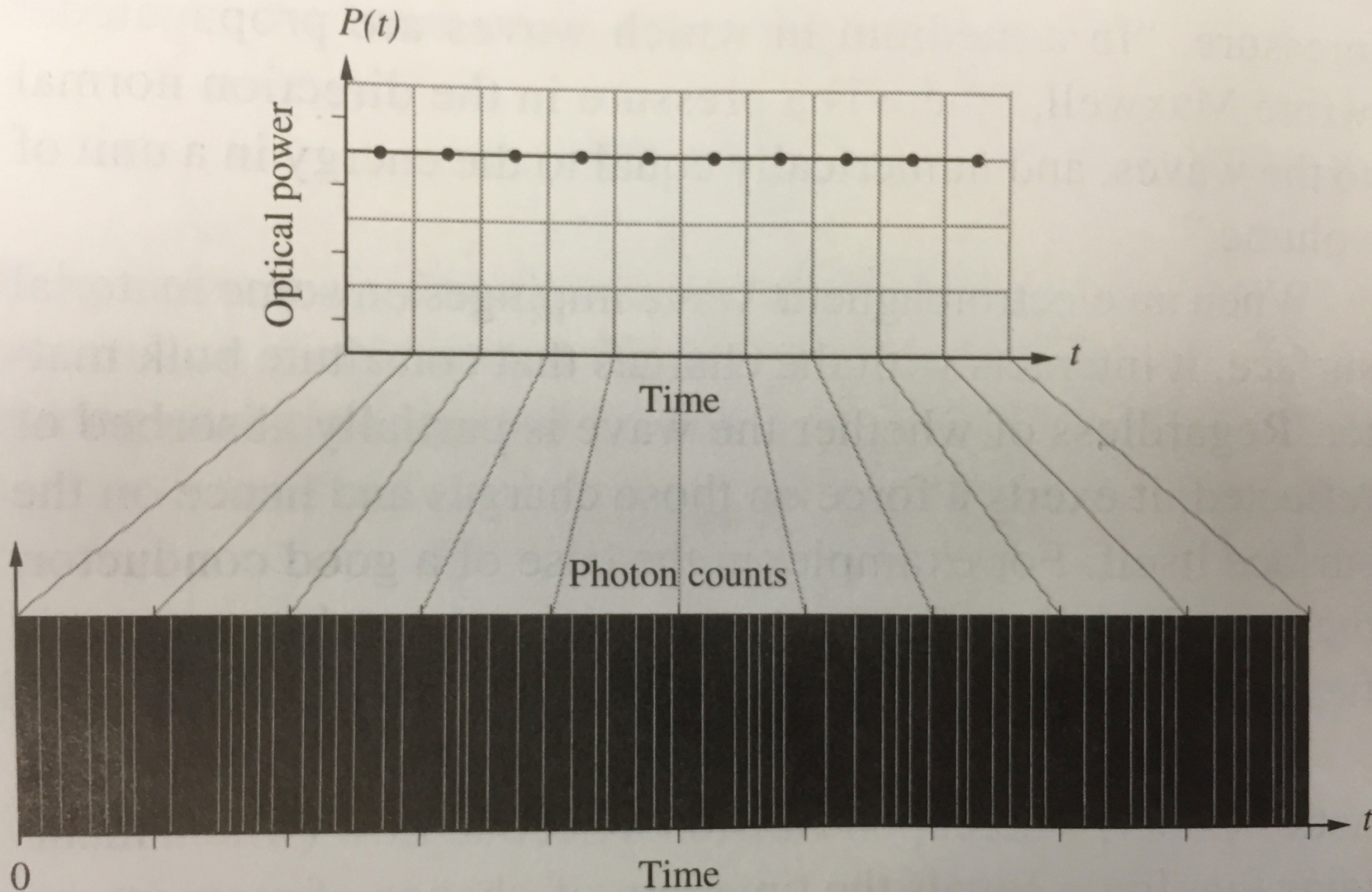


TABLE 3.1 The Mean Photon Flux Density for a Sampling of Common Sources

Light Source	Mean Photon Flux Density Φ/A in units of (photons/s·m ²)
Laserbeam (10 mW, He-Ne, focused to 20 μ m)	10^{26}
Laserbeam (1 mW, He-Ne)	10^{21}
Bright sunlight	10^{18}
Indoor light level	10^{16}
Twilight	10^{14}
Moonlight	10^{12}
Starlight	10^{10}

Photon Counts of a coherent light



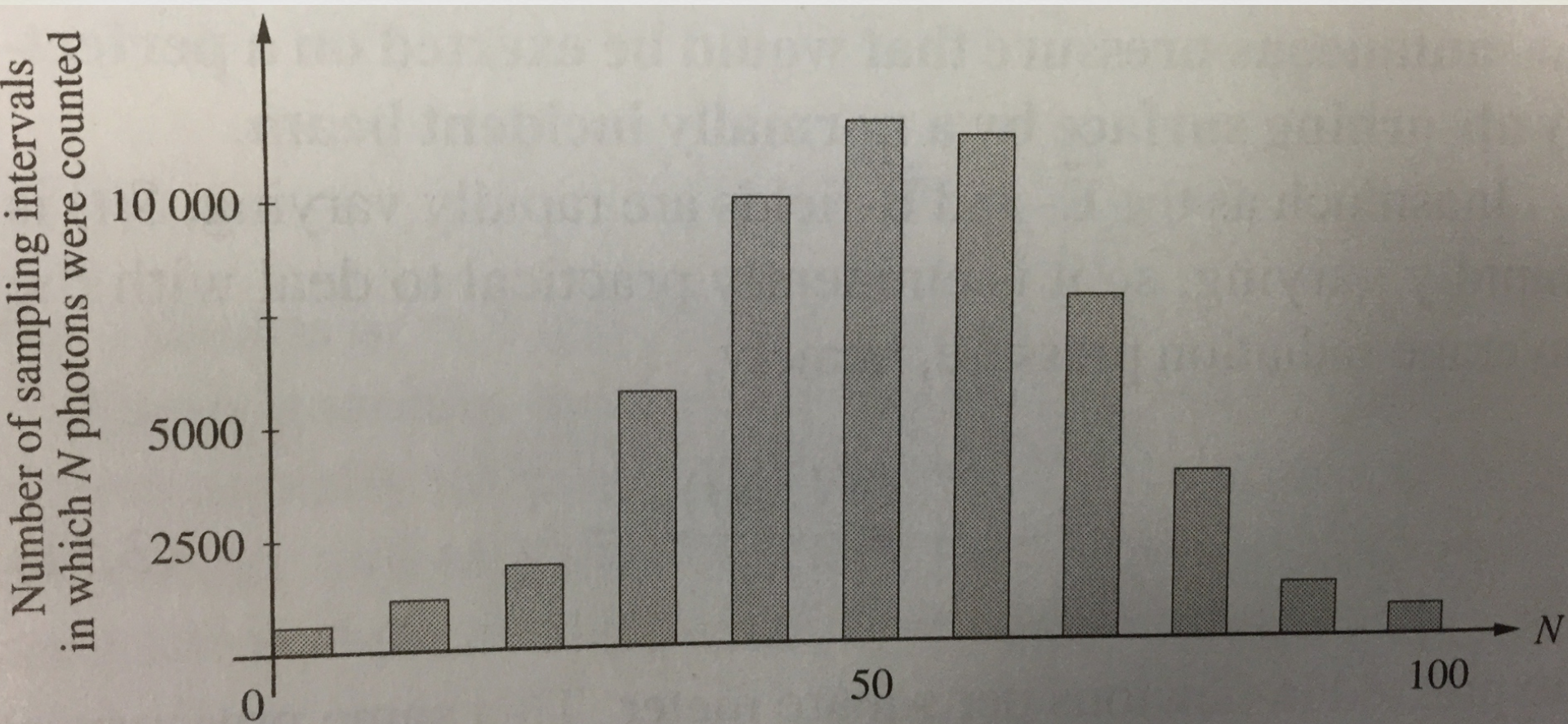


Figure 3.23 A typical histogram showing the probability or photon-count distribution for a beam of constant irradiance.

EM Wave VS Photon



$$\vec{E} = \vec{E}_0(\vec{r}, t) \cos(\vec{k} \cdot \vec{r} - \omega t) = \text{Re}\{\vec{E}_0(\vec{r}, t) e^{i(\vec{k} \cdot \vec{r} - \omega t)}\}$$

$$\langle u \rangle_T = \frac{1}{2} \epsilon_0 E_0^2(\vec{r}, t) = N(\vec{r}, t) \hbar \omega$$

$$N(\vec{r}, t) = \frac{\epsilon_0 E_0^2(\vec{r}, t)}{2 \hbar \omega}$$

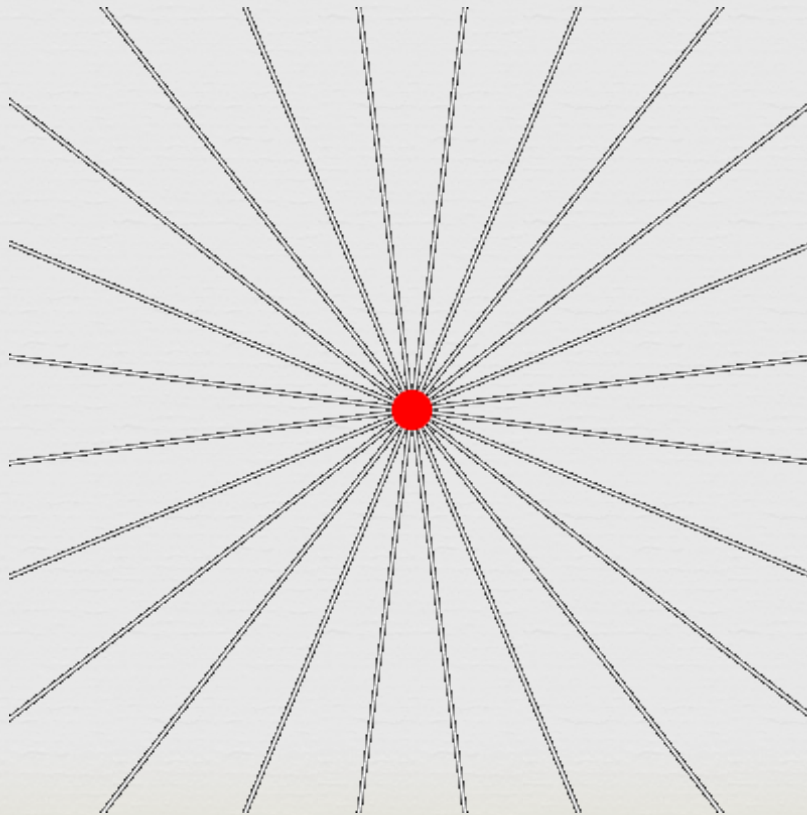
$$\text{Photon \# density} \sim E_0^2(\vec{r}, t)$$

Read textbook

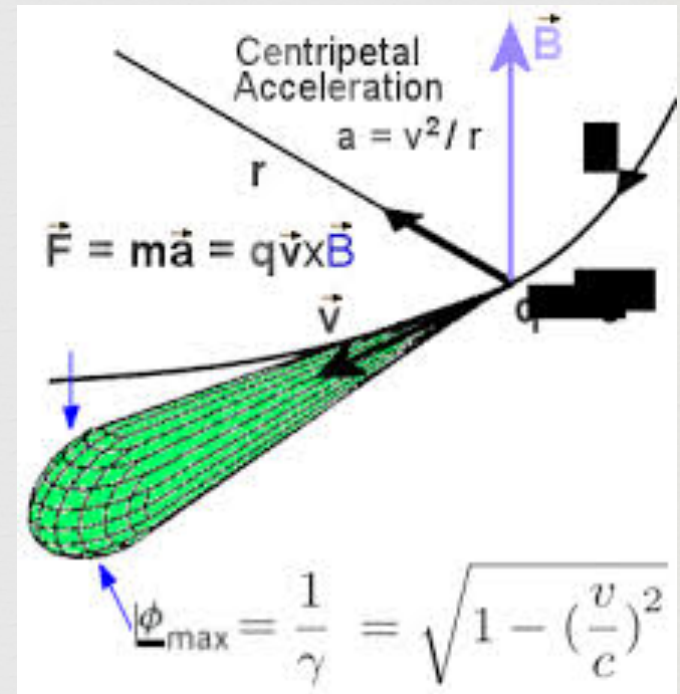
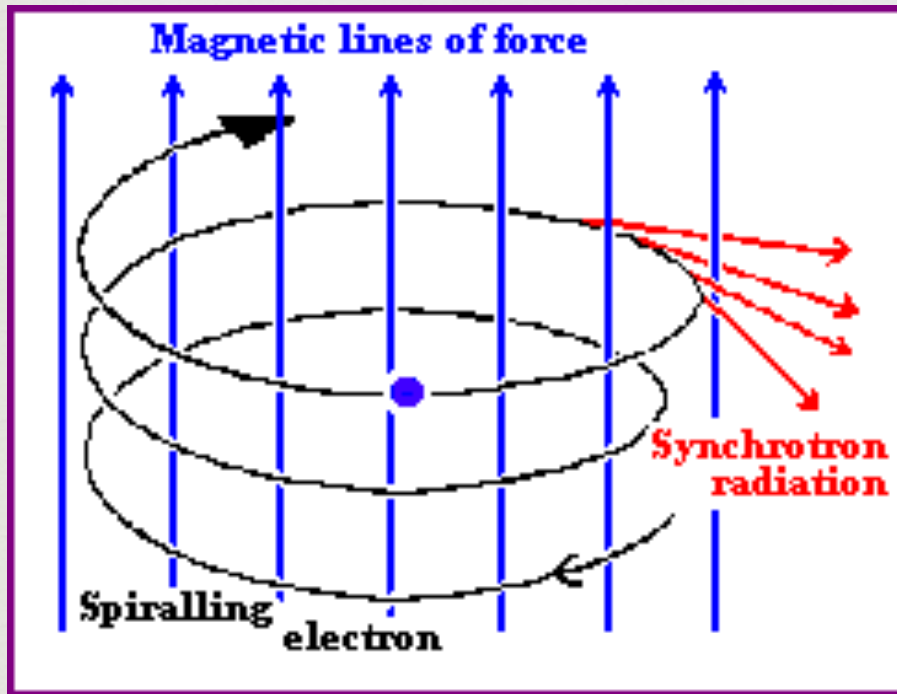
3.4 Radiation: Classical EM



⌘ Accelerating charge $\rightarrow$ Radiation



Synchrotron Radiation

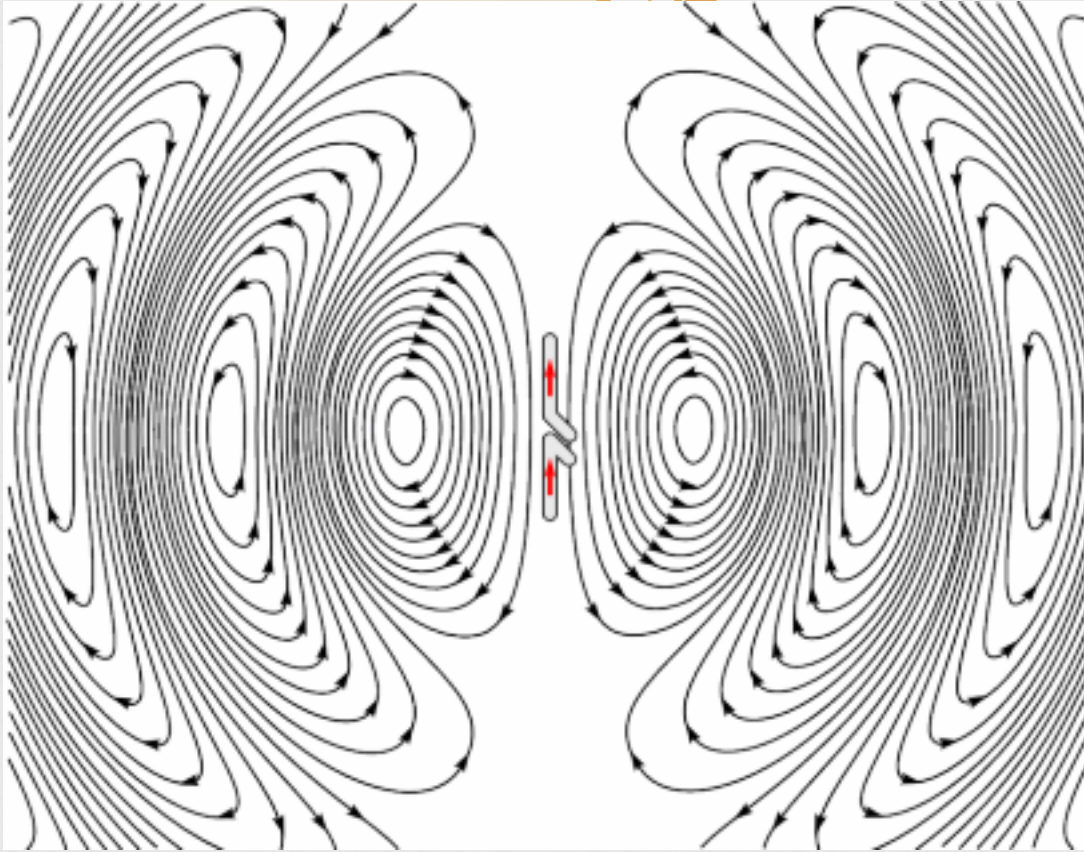


Synchrotron Light Source



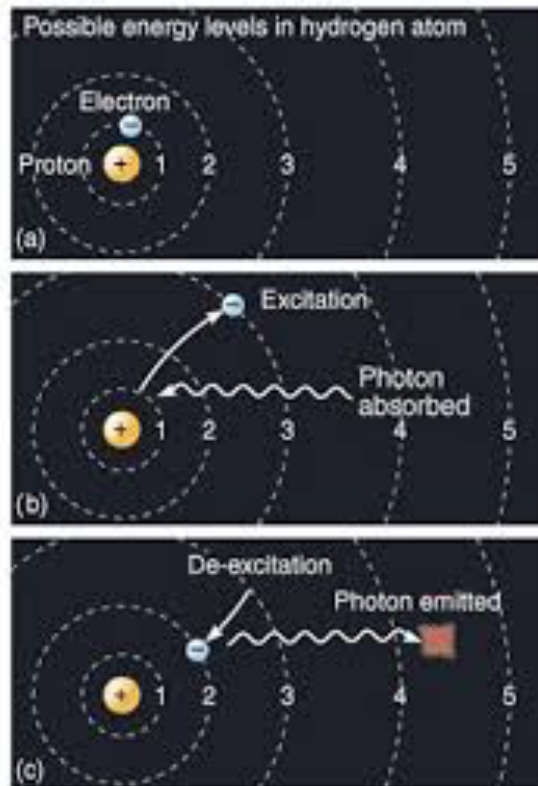
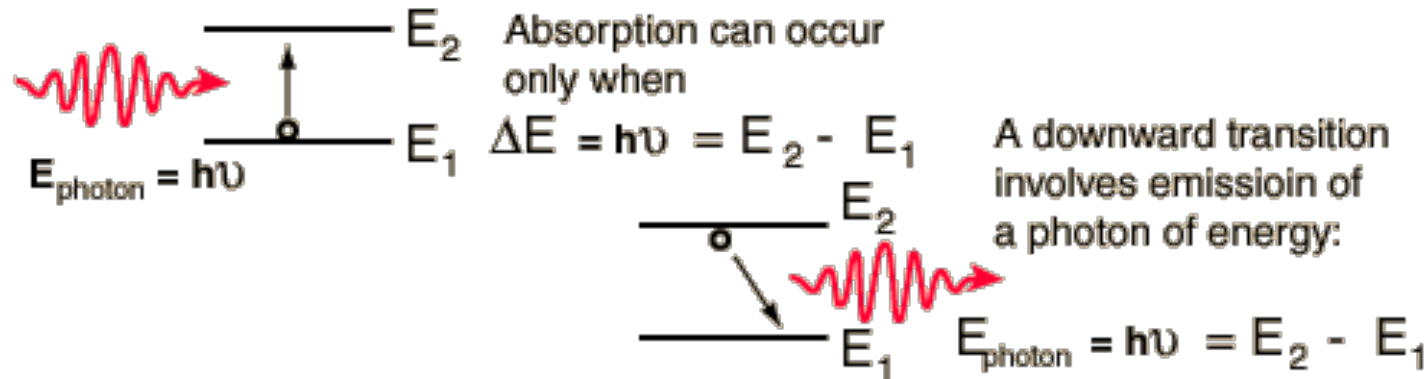
https://en.wikipedia.org/wiki/Synchrotron_light_source

Electric Dipole Radiation

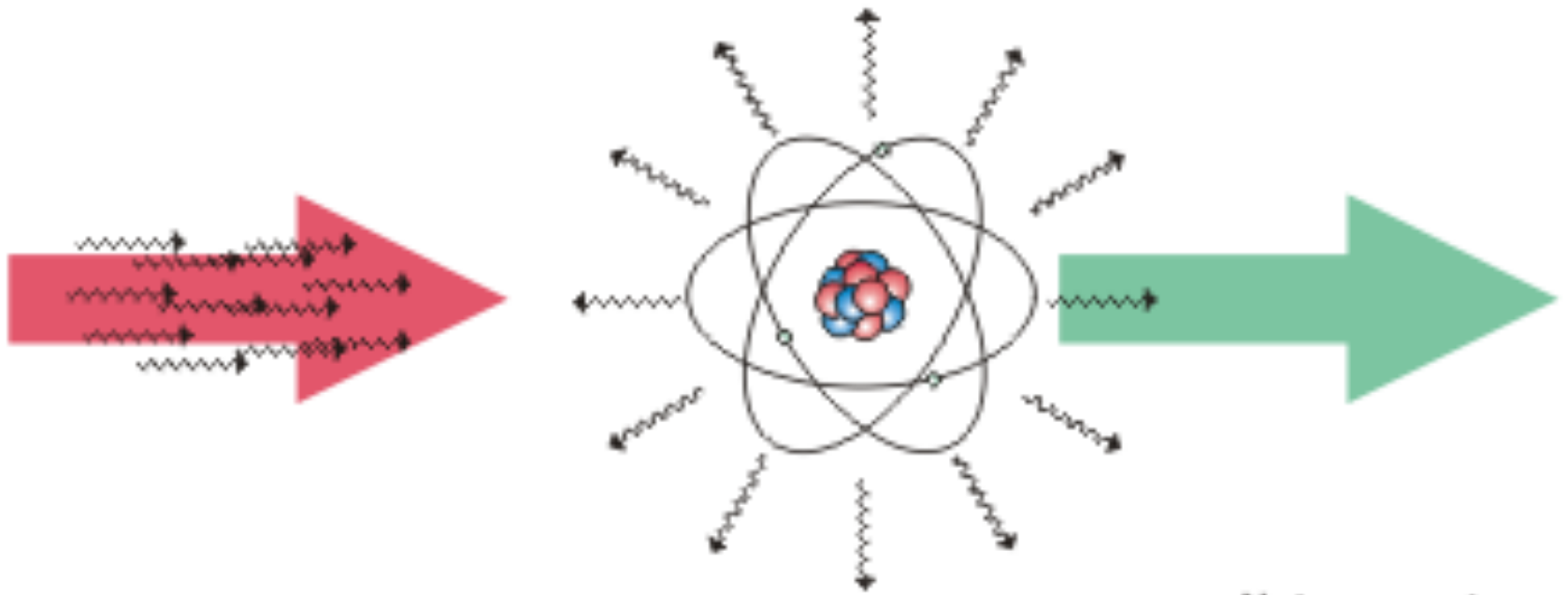


More on: https://en.wikipedia.org/wiki/Dipole_antenna

Quantum Physics of Radiation



Laser Cooling and Trapping



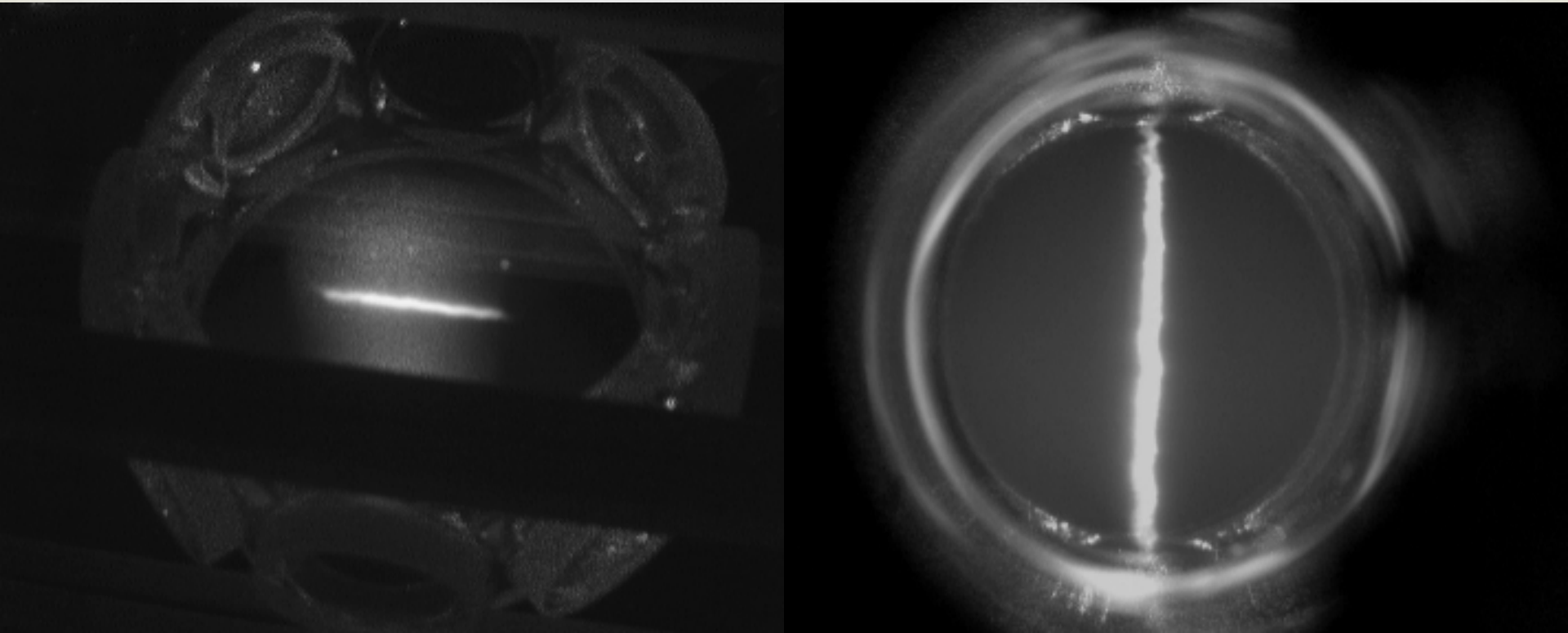
Incident photons absorbed:
momentum transfer = $\hbar k$

Spontaneous emission: total
momentum transfer = 0

Net momentum
transfer to atom in
direction of laser

1997 Nobel Prize in Physics

2D ^{85}Rb MOT @ HKUST



Number of atoms: $\sim 10^8$
Temperature: ~ 10 μK

Light in Bulk (Dielectric) Matter



$$\vec{D} = \epsilon_0 \vec{E} + \vec{P} = \epsilon \vec{E}$$

$$\vec{H} = \frac{\vec{B}}{\mu_0} - \vec{M} = \frac{\vec{B}}{\mu}$$

$$\epsilon_0 \Rightarrow \epsilon = \epsilon(\omega)$$

Dispersion

$$\mu_0 \Rightarrow \mu = \mu(\omega)$$

For most material (nonmagnetic) $\mu = \mu(\omega) \cong \mu_0$

$$\nabla \times \vec{E} = -\frac{\partial \vec{B}}{\partial t}$$

$$\nabla \times \vec{H} = \vec{J}_f + \frac{\partial \vec{D}}{\partial t}$$

$$\nabla \cdot \vec{D} = \rho_f$$

$$\nabla \cdot \vec{B} = 0$$

Dispersion (Dielectric)

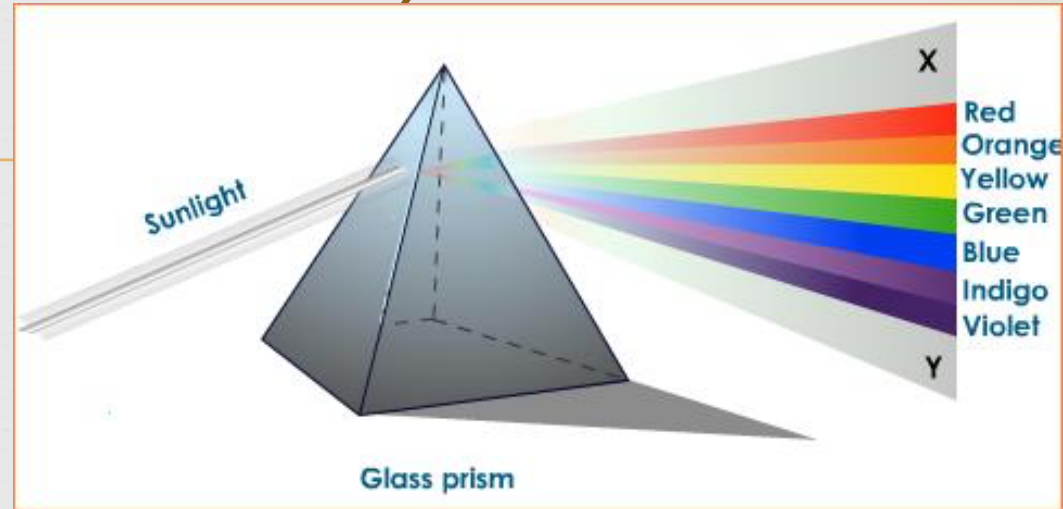
$$v(\omega) = \frac{1}{\sqrt{\epsilon(\omega)\mu(\omega)}} = \frac{c}{n(\omega)}$$

$$n(\omega) = \sqrt{\frac{\epsilon(\omega)\mu(\omega)}{\epsilon_0\mu_0}} = \sqrt{K_E(\omega)K_M(\omega)}$$

$$K_E(\omega) = \sqrt{\frac{\epsilon(\omega)}{\epsilon_0}}$$

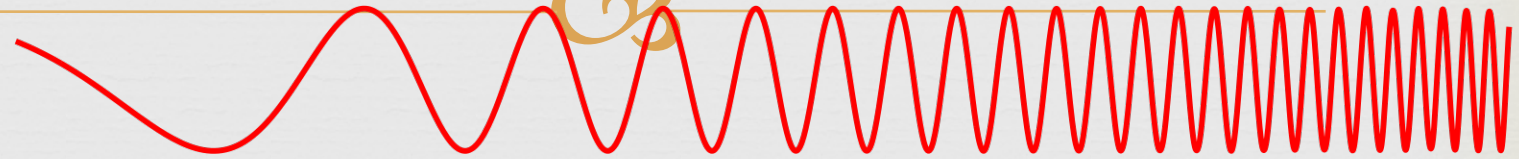
Dielectric constant

$$K_M(\omega) = \sqrt{\frac{\mu(\omega)}{\mu_0}} \cong 1$$



3.6 EM-Photon Spectrum

Penetrates Earth's Atmosphere?



Radiation Type
Wavelength (m)

Radio
 10^3

Microwave
 10^{-2}

Infrared
 10^{-5}

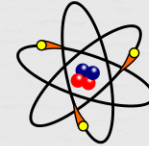
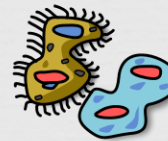
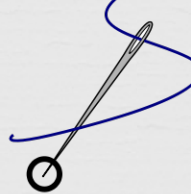
Visible
 0.5×10^{-6}

Ultraviolet
 10^{-8}

X-ray
 10^{-10}

Gamma ray
 10^{-12}

Approximate Scale
of Wavelength



Buildings

Humans

Butterflies

Needle Point

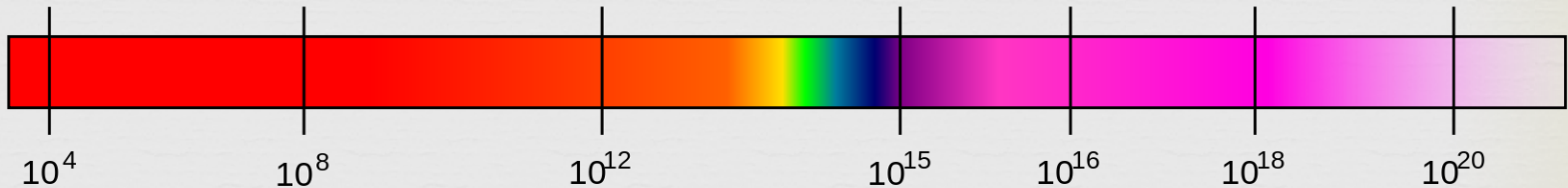
Protozoans

Molecules

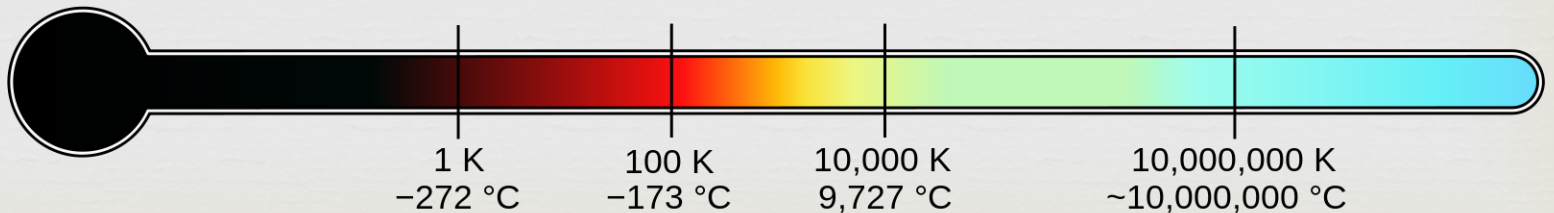
Atoms

Atomic Nuclei

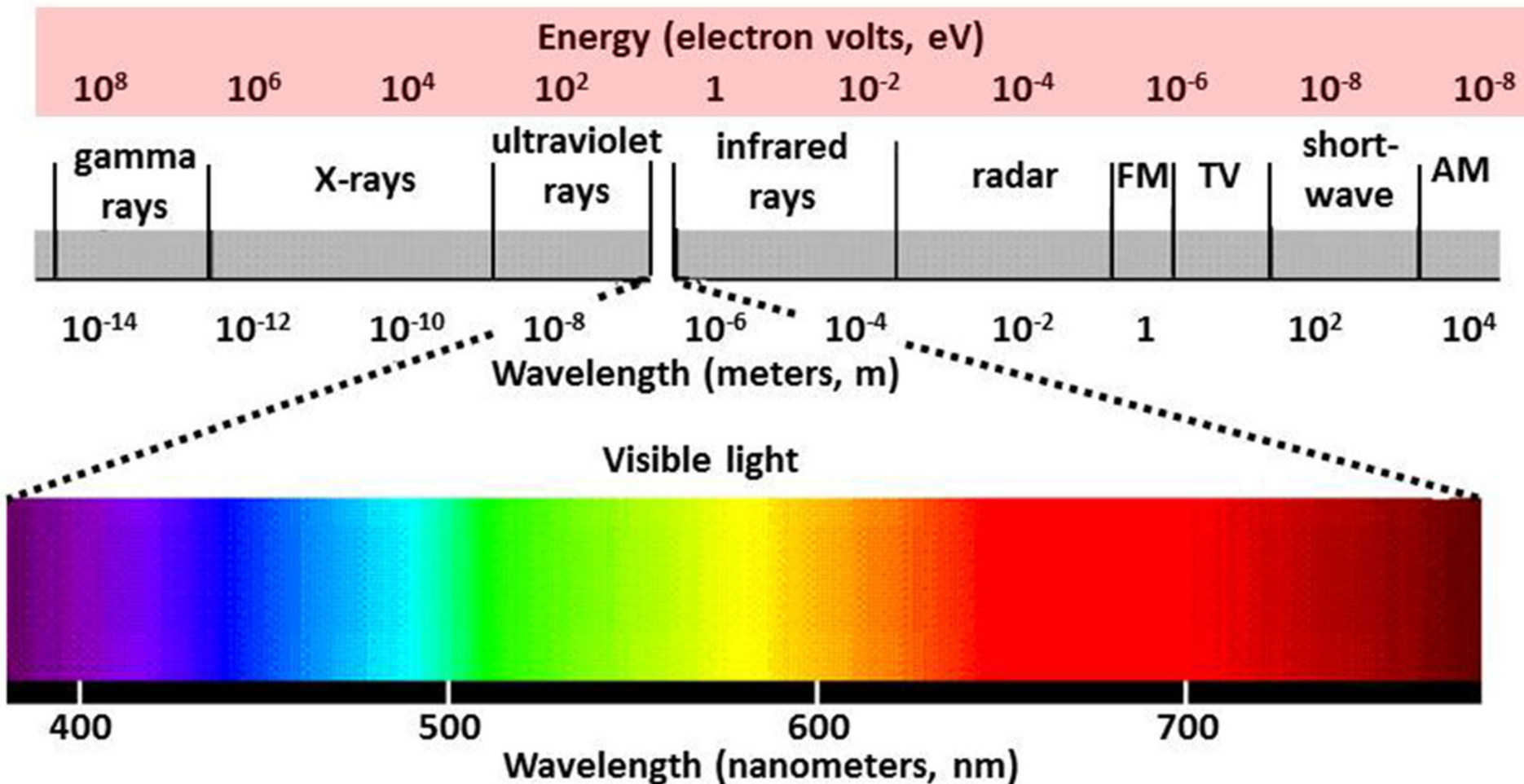
Frequency (Hz)



Temperature of
objects at which
this radiation is the
most intense
wavelength emitted



EM Spectrum



3.7 Quantum Field Theory



QFT: wave-particle duality

Wave is the field of its particle

Particle of the particle of its field.

QED (Quantum Electrodynamics): Quantum theory of EM field and light-matter interaction.

